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The structure and solubilization properties of aqueous solutions of lithium and sodium dodecyl sulfates

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Abstract

Micellization of lithium and sodium dodecyl sulfates (LDS and SDS respectively) was studied in aqueous solutions by conductometry. Besides the CMC inflection point the other one was observed on electrolytic conductivity plots. Phenomenological explanation of this event was proposed basing on overlapping the diffusion regions of electrical double layers. Solubilization properties of SDS and LDS solutions were determined before and after micelle formation.